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EVALUATION OF THE STAINING POTENTIAL OF SILVER DIAMINE FLUORIDE AND NANO SILVER FLUORIDE ON PRIMARY TEETH: AN IN VITRO STUDY

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ABSTRACT

Background: Despite the extensive literature validating the efficacy of Nanosilver fluoride (NSF) as a cariostatic agent, limited information exists regarding its surface staining potential. Aim: In vitro evaluation of surface staining changes over time of demineralized enamel of primary teeth treated with silver diamine fluoride (SDF), SDF and potassium iodide (KI), or NSF using VITA Easy Shade digital spectrophotometer, before treatment, one hour, two- and four weeks after treatment. **Results:** spectrophotometric color analysis showed that SDF-treated demineralized enamel exhibited the highest staining compared to either NSF or SDF+KI at the 3 time intervals and the results were statistically significant (p < 0.001). Comparing the staining potential between the SDF+KI and NSF groups, similar staining values were observed one hour after application. However, Δ E2, Δ L2, Δ E3, and Δ L3 values indicated the less staining potential of NSF on enamel surfaces after 2 and 4 weeks of application, however, results were only statistically significant for Δ L2 (p= 0.005). **Conclusion:** NSF could be a more esthetic alternative to SDF, and SDF+KI when applied to demineralized enamel of primary teeth.

KEYWORDS: Silver diamine fluoride, Nano silver fluoride, Potassium iodide, Primary teeth.

INTRODUCTION

Dental caries is one of the most common chronic diseases ⁽¹⁾. **Recently, caries management has shifted to** minimal intervention dentistry (MID) which offers a more conservative treatment by

managing and preventing carious lesions in their early stages without removing hard dental tissues.

MID frequently uses fluoridated and antimicrobial agents. The caries-arresting properties of silver diamine fluoride (SDF) have gained considerable

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attention. It is effective against cariogenic bacteria and is capable of remineralizing dentin and enamel⁽²⁾. However, the black stains that remain after application to carious dental tissues influence patients' and parents' acceptance of this treatment⁽³⁾. Aiming to overcome the black discoloration of SDF, potassium iodide (KI) was used in conjunction with it⁽⁴⁾. However, the arrested enamel lesions treated with this combination still showed perceptible discoloration⁽⁵⁾.

The recently introduced Nano silver compounds, such as Nano silver fluoride (NSF), are proposed to be more aesthetic cariostatic agents. The NSF application is safe for both adults and children to use, noninvasive, nontoxic, and reasonably priced ⁽⁶⁾. It is claimed to cause less discoloration or irritation to the soft tissues than SDF. It possesses the antibacterial actions of Silver Nanoparticles (AgNPs) and remineralizing properties of fluoride ⁽⁷⁻⁹⁾.

Despite the extensive literature validating the efficacy of Nanosilver fluoride (NSF) as a cariostatic agent⁽⁸⁻¹¹⁾, very limited information exists regarding its surface staining potential. Asghar et al. (2023) ⁽¹²⁾ systematically reviewed the literature regarding the efficacy of different approaches for minimizing SDF– mediated tooth discoloration, they found only one study that was performed by Espíndola et al. (2020) ⁽⁷⁾ which assessed the staining potential of NSF on dentin of permanent teeth.

Accordingly, and up to our knowledge, this is the first in-vitro study comparing the staining effect of SDF, SDF+KI, and NSF on demineralized enamel of primary teeth using VITA Easy Shade digital spectrophotometer, before treatment, one hour, twoand four weeks after treatment.

MATERIALS AND METHODS

An in vitro study was designed to evaluate the staining potential of SDF, SDF+KI, and NSF on primary teeth after one hour, two weeks, and four weeks of application of these cariostatic agents. The study was approved by the Ethics Committee of the Faculty of Dentistry, Minia University (Research code: 558/2022).

Thirty extracted sound human primary teeth were collected from the outpatient clinic of the pediatric and community dentistry department at Minia University Dental Hospital. Extraction was indicated for one of the following reasons: prolonged retention or orthodontic purposes. The children's legal guardians permitted using their extracted teeth for research purposes. Sound primary teeth without cracks, fractures, caries, developmental anomalies, or restorations, were selected for the study.

Sample preparation

Selected teeth were cleaned with a prophylactic brush and washed with running water. Then, they were stored in distilled water and thymol until the time of the experiment. Circular adhesive tapes of 6 mm diameter were applied on the tooth surface. The remaining tooth surface was coated with an acidresistant nail varnish. After drying the nail varnish, the adhesive tapes were removed leaving a circular window without nail varnish.

Teeth immersion in a demineralizing solution containing 2.2mM CaCl2, 2.2mM KH2PO4, 0.05mM Acetic acid, and pH adjusted to 4.4 1M KOH ^(13,14) was performed over seven days. Then, each tooth was mounted separately on an acrylic block. After being numbered, the thirty samples were randomly divided into three groups (n = 10 per group). Group (1) teeth received SDF treatment (control group); group (2) teeth received (SDF+KI) treatment, and group (3) teeth received NSF treatment.

Cariostatic Agents

38% **SDF** (TOOTHMATE [™], Egypt) and 30% **KI** (TOOTHMATE [™], Egypt) were purchased while **NSF** in a liquid form (1000 ppm) was laboratory prepared (Nanotech for Photo-Electronics, Cairo - Egypt).

NSF preparation: Using a magnetic stirrer, chitosan (28.7 ml, 2.5 mg/ml) was first dissolved in 1% acetic acid overnight. The mixture was then put into a flask and transferred to an ice-cold bath after being filtered through a vacuum filter unit. Then the mixture was stirred vigorously before adding 1 ml of silver nitrate (0.11 mol/L). Afterwards, a drop-by-drop, freshly prepared sodium borohydride (0.3ml, 0.8mol/L) was added. Silver (Ag+) was reduced instantly and the solution turned from colorless to pale yellow and finally dark yellow. Sodium fluoride (10,147 parts per million) was added, after taking the flask out of the ice bath⁽⁷⁾. The prepared NSF was characterized using a Transmission Electron Microscope (Figure 1).

Cariostatic agent application

Agents were applied onto the exposed demineralized enamel lesions with a micro brush. Group (1) teeth were treated with 38% SDF for two minutes before rinsing with distilled water. Group (2) teeth were treated with SDF and KI right after the SDF was applied. KI was applied with a microbrush to the same location until the reactionary white precipitate became transparent. Using a microbrush, each tooth in group (3) received two drops of NSF solution applied to the exposed enamel window. After two minutes of contact with the solution, the tooth surface was rinsed with distilled water.

Color Measurement

Teeth color was measured using a digital spectrophotometer (VITA Easyshade, VITA Zahnfabrik, Bad Säckingen, Germany) at 4 different time points (before agent application, then 1 hour, 2- and 4 weeks after application). Tooth staining analyzation was done based on the CIELAB color space and expressed as ΔE and ΔL . Prior to each color measurement, calibration of the spectrophotometer was performed ⁽¹⁵⁻¹⁷⁾.

The dental staining was determined using the CIELAB system parameters, which are (L*a*b*), where L* stands for lightness and the mean value ranges from 0 (black) to 100 (white). The values of a* and b*, respectively, represent the red-green and yellow-blue axes. The total color differences (ΔE) of enamel staining were computed using the following formula^(7,18,19). $\Delta E = ((\Delta L)^2 + (\Delta a)^2 + (\Delta b)^2)^{1/2}$.

In each separate group, the difference between baseline and after one hour ($\Delta E1$ and $\Delta L1$) were compared with those of baseline- two weeks ($\Delta E2$ and $\Delta L2$) and baseline- four weeks ($\Delta E3$ and $\Delta L3$).

Statistical analysis

SPSS (Statistical Package for the Social Science, version 26, IBM, and Armonk, New York) was used to analyze the collected data. To determine data compliance with normal distribution the Shapiro test was used. Quantitative data were expressed as mean

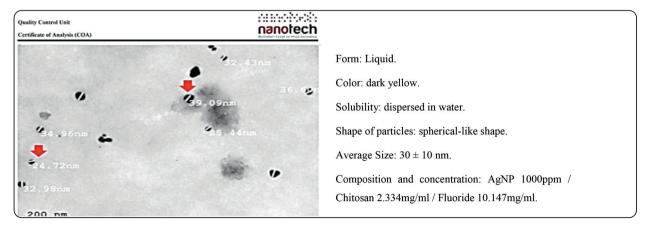


Fig. (1) Transmission Electron Microscope micrograph of NSF solution immediately after preparation. It shows predominantly spherical silver nanoparticles (arrow) dispersed in the colloidal solution.

and standard deviation (SD) and compared with the ANOVA test followed by post-hoc analysis. In each group, the difference between baseline and after one hour (Δ E1 and Δ L1) were compared with those of baseline-two weeks (Δ E2 and Δ L2) and baseline-four weeks (Δ E3 and Δ L3) by paired t-test. The confidence level was kept at 95%; hence, p-value <0.05 was considered significant.

RESULTS

In each group, ΔL values indicated that enamel lesions got darker with time. A similar finding was observed for ΔE values for SDF and NSF groups while for the SDF+KI treated lesions, $\Delta E2$ was higher than $\Delta E3$.

SDF-treated demineralized enamel exhibited the highest staining compared to either NSF or SDF+KI at the 3 time intervals and the results were statistically significant (p < 0.001).

Comparing the staining potential between the SDF+KI and NSF groups, similar staining values were observed one hour after application. However, $\Delta E2$, $\Delta L2$, $\Delta E3$, and $\Delta L3$ values indicated less staining effect of NSF on enamel surfaces after 2 and 4 weeks of application, compared to SDF+KI. The results were only statistically significant for $\Delta L2$ (p= 0.005) (Table: 1, Figure 2).

TABLE (1) Enamel color changes of ΔE and ΔL after 1 hour, 2 weeks and 4 weeks of application of cariostatic agents in the three groups

	ΔΕ1	ΔL1	ΔΕ2	ΔL2	ΔΕ3	ΔL3
	(Baseline-after	(Baseline-after	(Baseline-two	(Baseline-two	(Baseline-four	(Baseline-four
	one hour)	one hour)	weeks)	weeks)	weeks)	weeks)
SDF	51.76 ± 19.10	-52.76 ± 12.65	68.21 ± 8.07	-62.83 ± 8.47	73.22 ± 9.94	-67.57 ± 10.36
SDF& KI	12.60 ± 14.70	-2.19 ± 5.90	30.31 ± 23.59	-22.73 ± 16.43	28.65 ± 17.39	-27.78 ± 16.94
NSF	11.19 ± 17.14	-4.08 ± 6.09	15.50 ± 16.15	-7.44 ± 5.14	23.23 ± 17.26	-16.45 ± 11.13
p value	< 0.001*	< 0.001*	< 0.001*	< 0.001*	< 0.001*	< 0.001*
Post-hoc						
p1 value	< 0.001*	< 0.001*	< 0.001*	< 0.001*	< 0.001*	< 0.001*
p2 value	< 0.001*	< 0.001*	< 0.001*	< 0.001*	< 0.001*	< 0.001*
p3 value	0.855	0.635	0.064	0.005*	0.434	0.064

Data expressed as mean (SD). p-value was significant (*) if < 0.05. p-value compared between the three agents' groups, p1 value between SDF vs. SDF& KI, p2 value between SDF vs. NSF, and p3 value between NSF vs. SDF& KI.

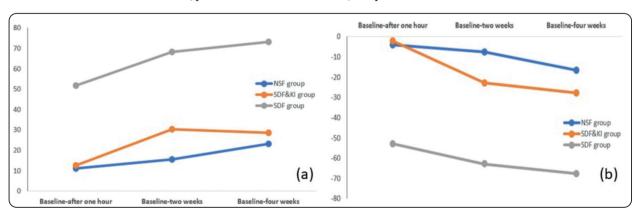


Fig. (2): (a) ΔE and, (b) ΔL enamel staining of the three groups at different time intervals.

DISCUSSION

Recent caries management emphasizes non-restorative, non-invasive, and micro-invasive therapeutic approaches that successfully cease the caries progression at the lesion level and minimize the loss of healthy tooth structure.

SDF represents a promising advancement in caries management. Despite its proven efficacy as a cariostatic agent, the resulting black stains limit its use in the anterior teeth and decrease parent and patient acceptance.

Recently, NSF was introduced as a cariostatic agent with reduced staining potential ⁽⁹⁾. Thus, the current study was performed as an in vitro study comparing the staining potential of SDF, SDF+ KI and SNF on demineralized primary teeth enamel.

The current study was designed as a laboratory study since this study type permits the comparison between different interventions under standardized conditions⁽²⁰⁾. Primary teeth were selected as cariostatic agents are commonly used to arrest caries in the primary teeth of children with decreased cooperation⁽²³⁾. Moreover, the effectiveness of cariostatic agents in primary dentition has been confirmed as the enamel is thinner with greater permeability that is about 150 times greater than that in permanent teeth allowing for fluoride diffusion⁽²⁴⁾.

Several studies have tested the staining potential of cariostatic agents using either bovine teeth⁽²¹⁾ or permanent human teeth ^(3,7,13,22), however, they differ in structure and morphology compared to primary teeth. Also, cariostatic agents were tested on demineralized enamel rather than dentine since cariostatic agents show great success when applied to early carious lesions affecting enamel.

Samples were immersed in a demineralizing solution for seven days to produce simulated demineralized enamel lesions. This complies with Gupte et al. (2021)⁽¹³⁾ and El-Desouky et al. (2022)⁽¹⁴⁾ who followed the same demineralization protocol. Then specimens were randomly allocated into three groups with three different treatment approaches^(13,14).

The present study investigated the change in color in the three groups at four-time points (before treatment and one hour, two- and four-weeks after treatment) which complies with the time intervals of Nguyen et al. $(2017)^{(25)}$, Espíndola et al. $(2020)^{(7)}$ and Gupte et al. $(2021)^{(13)}$.

Color change was assessed using a digital spectrophotometer as it is amongst the most accurate instruments for color analysis in dentistry. Accurate tooth color measurements can be obtained with a digital spectrophotometer, which is superior to visual analysis, that is subject to observer variability. Additionally, human physiological variables (such as age, emotional state, and fatigue) and lighting conditions can impact the assessment of visual color⁽²⁷⁾. In the current study, spectrophotometric assessment was performed according to CIE L*a*b* color analysis to calculate ΔE and ΔL similar to studies assessing tooth color variations either after bleaching or cariostatic agent application in the permanent dentition ^(7, 13, 25, 26).

In the current study, spectrophotometric results showed that ΔE was increased in the three groups after the cariostatic agents' application at the three-time intervals. Clinically, a color change is visible when ΔE values are higher than 3.7 units ^(5, 26). $\Delta E1$, $\Delta E 2$ and $\Delta E 3$ values of the three groups exceeded the critical value of clinical detection (3.7 units).

The current study revealed that SDF exhibited the highest staining potential at the 3 time intervals and the results were found to be statistically significant (p < 0.001). Staining was reduced by the application of KI. This goes with the results of Patel et al. (2018)⁽³⁾ and Gupte et al. (2021)⁽¹³⁾ who concluded that SDF can visibly stain dental hard tissues and

that the application of KI following SDF helps in reducing the staining. However, Li et al (2016)⁽⁴⁾ reported that SDF+KI-treated carious lesions were stained similarly to applying SDF alone after 30 months. This difference may be attributed to studying the staining effect on root caries in permanent teeth of elderly patients. Also, the results might be affected by the oral environment as well as the extended follow-up period ^(3, 4, 13).

On the other hand, minimal staining of the enamel of teeth treated by NSF was observed. The staining of the NSF was similar to SDF+KI one hour after application. However, $\Delta E2$, $\Delta L2$, $\Delta E3$, and $\Delta L3$ values indicated less staining effect of NSF on enamel surfaces after 2 and 4 weeks of application, compared to SDF+KI. The results were only statistically significant for $\Delta L2$ (p=0.005). The results after 2 weeks coincide with the results of the in vitro study of Espíndola et al. (2020)⁽⁷⁾ who compared similar cariostatic agents on dentin staining of permanent molars. However, Espíndola et al. (2020)⁽⁷⁾ reported an enhancement of the values of ΔE and ΔL after 4 weeks of NSF application. This may be attributed to applying the agents to dentin rather than enamel and the brushing cycles that seemed to remove the stained layer ⁽⁷⁾.

The limitation of the current study was the assessment of color change under in-vitro conditions. Therefore, further clinical studies are recommended to evaluate the staining effect of NSF on primary teeth under the challenging oral environment over an extended period and to test different concentrations of NSF in vivo to study the cariostatic effect and staining potential on primary teeth.

CONCLUSION

Within the limitations of the current in vitro study, NSF could be a more esthetic alternative to SDF, and SDF+KI when applied to demineralized enamel of primary teeth.

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